

Article

Characterization of a Novel Ginsenoside MT1 Produced by an Enzymatic Transrhamnosylation of Protopanaxatriol-Type Ginsenosides Re

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Abstract: Background: Ginsenosides, triterpene saponins of *Panax* species, are considered the main active ingredients responsible for various pharmacological activities. Herein, a new protopanaxatriol-type ginsenoside called “ginsenoside MT1” is described; it was accidentally found among the enzymatic conversion products of ginsenoside Re. Method: We analyzed the conversion mechanism and found that recombinant β -glucosidase (MT619) transglycosylated the outer rhamnopyranoside of Re at the C-6 position to glucopyranoside at C-20. The production of MT1 by trans-rhamnosylation was optimized and pure MT1 was obtained through various chromatographic processes. Results: The structure of MT1 was elucidated based on spectral data: (20S)-3 β ,6 α ,12 β ,20-tetrahydroxydammarane-20-O- $[\alpha$ -L-rhamnopyranosyl(1 \rightarrow 2)- β -D-glucopyranoside]. This dammarane-type triterpene saponin was confirmed as a novel compound. Conclusion: Based on the functions of ginsenosides with similar structures, we believe that this ginsenoside MT1 may have great potential in the development of nutraceutical, pharmaceutical or cosmeceutical products.

Keywords: ginsenosides; dammarane-type triterpene saponin; ginsenoside MT1; transglycosylation; biotransformation; biotechnology

1. Introduction

Ginsenosides—triterpene saponins composed of a dammarane skeleton with several glycosylation positions—are generally considered the main active components of ginseng, that has been used as a traditional herbal medicine in East Asian countries for thousands of years to stimulate physical and mental activity [1,2]. Ginsenosides can be categorized as protopanaxadiol (PPD) and protopanaxatriol (PPT) saponins based on the structure of their aglycon. More than 100 kinds of ginsenosides have been found [3,4]; their pharmacological effects vary according to their attached sugars and aglycon structures [5–7]. Six major kinds of ginsenoside (Rb₁, Rb₂, Rc, Re, Rd and Rg₁), as shown in Figure S1, constitute >90% of the total ginsenosides in ginseng [2,8]. They are relatively abundant in ginseng and can be converted into minor ginsenosides, which naturally exist in smaller amounts and have higher chemical reactivities [9–11].

Various methods have been reported for preparing minor ginsenosides with high conversion efficiency and few byproducts [12]. Many enzymes have been explored to efficiently convert major ginsenosides into pharmacologically active, rare minor ginsenosides [5,13,14]. Prior to this study, our research group cloned, characterized and applied many efficient glycoside hydrolases for the gram-scale production and purification of minor ginsenosides [15–21].

In our investigation into the enzymatic conversion of major ginsenosides, a new transformation product of ginsenoside Re that is different from the other hydrolyzing products (ginsenoside PPT, F₁, Rg₁ and Rg₂) was accidentally found using a novel recombinant glucosidase (MT619) [15]. This compound was prepared at the gram scale using recombinant MT619 and isolated from the product mixture using chromatographic methods. Finally, the structure of this novel ginsenoside was elucidated based on spectral data.

2. Materials and Methods

2.1. Chemicals and Reagents

The standard forms of various ginsenosides (Rg₁ and Re) used in the present study were purchased from Sigma Co., Ltd. (Louis, MO, USA). F₁, Rg₂(S), Rh₁(S) and PPT were prepared as described in our previous study [22]. The other chemical reagents used were at least extra or better in quality than pure grade. Butanol, ethyl acetate, chloroform, ethanol and pyridine-d₅ (Merck KGaA, Darmstadt, Germany) were purchased from Sam Chun Pure Chemical Co. (Pyeongtaek, Korea).

2.2. Biotransformation of Ginsenosides Using Recombinant MT619

Recombinant MT619 was expressed in *Escherichia coli* BL21 as described previously [15]. Briefly, the MT619-harboring pEX vector was inserted into the *E. coli* BL21 strain using heat-shock transformation. The cells grown in LB medium were supplemented with ampicillin at 37 °C until the culture reached an optical density of 600 nm (OD₆₀₀) of 0.6, at which point protein expression was induced by adding 0.1 mM isopropyl-β-D-thiogalactopyranoside (IPTG). The bacterial cells were incubated further for 24 h at 18 °C and then harvested by centrifugation at 4000 rpm for 15 min. The cells were washed with 50 mM sodium phosphate and then resuspended in 50 mM sodium phosphate and 1% Triton X-100 (pH 7.0). The cells were disrupted by ultrasonication (Vibra-Cell; Sonics & Materials, Danbury, CT, USA). Intact cells and debris were removed by centrifugation at 13,000 rpm for 10 min to obtain a crude cell extract. His-tagged MT619 protein was purified by a HisTrap column (GE Healthcare, Menlo Park, California, USA). The 100 mM imidazole eluted protein was further purified by DEAE-cellulose DE-52 chromatography (Whatman, Maidstone, UK). Purified MT619 was used to examine its MT1 production efficiency by reacting with ginsenoside Re (2.0 mg/mL, pH 7.0) in a shaking incubator at 37 °C for 24 h. The ginsenosides in the samples were extracted with an equal volume of butanol and identified by thin layer chromatography (TLC).

2.3. Optimization of the Substrate Concentration

Ginsenoside MT1 production was evaluated using crude MT619 cell extracts. To determine the optimal concentration of selected substrates for the biotransformation reaction, cells were mixed with an equal volume of a substrate at 3.0–22.5 mg/mL at 37 °C. Samples were then withdrawn at regular intervals and analyzed by TLC.

2.4. Scaled-Up Ginsenoside MT1 Production

Production was scaled up to 1 L and a final concentration of 6.0 mg/mL substrate (ginsenoside Re). The transformation was performed by adding 300 mL of crude cell lysate in a shaking incubator at 200 rpm and 37 °C. After incubating for 24 h, the mixture was centrifuged at 4000 × *g* for 15 min and the supernatant was loaded into a column packed with HP20 resin (120 g) (Sigma-Aldrich, St Louis, MO, USA). One liter of water was used to remove unbound hydrophilic compounds and free

sugar molecules, and the absorbed ginsenosides were eluted using three bed volumes of 95% ethanol. The eluted ethanol was evaporated in vacuo to produce dry powders.

2.5. Purification of MT1 from the Enzymatically Converted Products

Chromatographic pre-purification was performed with an LC-Forte/R system (YMC Korea Co. Ltd., Sungnam, Korea). One gram of the crude MT1 was dissolved in 20 mL of 10% methanol and centrifuged at $4000 \times g$ for 20 min. After filtering with a syringe filter (0.2 μm), the dissolved sample was subjected to liquid chromatography (YMC-C18, 25 μm 120 g, 39 mm \times 157 mm) and eluted with methanol-water (300 mL each at 5:5, 6:4, 7:3 and 8:2) to yield 13 fractions. The elution was fractionated every 100 mL, and the fractions containing MT1 were collected for purification.

2.6. Recycling Preparative High-Performance Liquid Chromatography (RPHPLC) Purification of MT1 from Biotransformed Products

The collected MT1 was further purified using RPHPLC (JAI NEXT Recycling Preparative HPLC LC-9210II NEXT, Japan Analytical Industry Co, Tokyo, Japan). RPHPLC was performed using a pre-packed column (JAIGEL-ODS-AP-L, 10 μm , 20 mm i.d. \times 500 mm) purchased from Japan Analytical Industry Co. (Japan). The mobile phase was 50% acetonitrile; the flow rate was 7.0 mL/min. The MT1 solution was prepared by dissolving a sufficient quantity of the collected MT1 in 50% acetonitrile to give a final concentration of 35 mg/mL; 10 mL of the MT1 solution was loaded for purification.

2.7. High-Performance Liquid Chromatography (HPLC) Analysis

HPLC analysis of samples in the present study was performed using an Agilent 1260 Infinity HPLC system (Agilent Co., Santa Clara, CA, USA). Ginsenosides were separated on an YMC ODS C18 column (5 μm , 250 \times 4.6 mm; YMC, Kyoto, Japan) with a guard column (Eclipse XDB C18, 5 μm , 12.5 \times 4.6 mm; Agilent Co., Santa Clara, CA, USA). The gradient elution system consisted of water (A) and acetonitrile (B), and used the following gradient program: 0 \rightarrow 10 min, 20% B; 10 \rightarrow 40 min, 20 \rightarrow 32% B; 40 \rightarrow 48 min, 32 \rightarrow 42% B; 48 \rightarrow 60 min, 42 \rightarrow 45% B; 60 \rightarrow 83 min, 45 \rightarrow 75% B; 83 \rightarrow 85 min, 75 \rightarrow 100% B; 85 \rightarrow 95 min, 100% B; 95 \rightarrow 95.01 min, 100 \rightarrow 20% B; 95.01 \rightarrow 100 min, 20% B. The detection wavelength was set to 203 nm and the flow rate was 1.6 mL/min.

2.8. Nuclear Magnetic Resonance (NMR) and Mass Spectrometry (MS) Analyses

NMR spectra were obtained using a Bruker AVANCE III 700 NMR spectrometer (Bruker, Rheinstetten, Germany) at 700 MHz (^1H) and 175 MHz (^{13}C), with chemical shifts given in ppm. Samples were separated with an Acquity UHPLC BEH C18 column (2.1 \times 100 mm i.d., 1.7 μm) at room temperature. The mobile phases comprised deionized water containing 0.2% (v/v) formic acid (A) and acetonitrile (B). Quantitative analysis of the novel compound was performed using an LTQ-Orbitrap XL mass spectrometer (Thermo Scientific, Bremen, Germany) equipped with an ESI source (Thermo Electron, Bremen, Germany) in negative ion mode with a mass range of m/z 100–1500 Da. Optimal conditions were employed as follows: capillary voltage 20 V, capillary temperature 350 $^\circ\text{C}$, spray voltage 3.5 kV and tube lens voltage 110 V.

2.9. Acidic Hydrolysis of MT1 and TLC Analysis

The purified compound (10 mg) was heated with 5% HCl mixture for 2 h at 70 $^\circ\text{C}$. The residues from filtration and standard sugars were compared through cellulose TLC (Butanol:Ethanol:H₂O, 50:50:30). TLC was conducted using 60F₂₅₄ silica gel plates (Merck, Darmstadt, Germany) and CHCl₃-CH₃OH-H₂O (65:35:10, v/v/v) as the developing solution. The results were visualized with 10% H₂SO₄ by heating at 110 $^\circ\text{C}$ for 5 min.

3. Results

3.1. Ginsenoside Re Transformation of MT619

MT619 was purified by Ni column and DEAE column chromatography as described in a previous study [15]. To analyze the transformation pathways, ginsenosides Re and Rg₂ were reacted as substrates with purified MT619 ginsenoside and F₁ and the conversion products were subjected to TLC analysis. The newly produced band (MT1), which had a different R_f value from those of ginsenoside Rg₁ and Rg₂, was identified from Re reaction mixtures (Figure 1). Rg₂(S) was also hydrolyzed by MT619 into PPT, but MT1 was not reported. Interestingly, MT1 was only found in the Rg₂ mixture containing F₁ (Sample 5 in Figure 1), which clearly suggests that F₁ is the recipient in trans-rhamnosyl reactions. No MT1 was formed in the PPT conversion mixture, suggesting that PPT cannot receive rhamnose. Rg₁ and Rh₁ were not found in the Re and Rg₂ conversion mixtures; the existence of rhamnose was inferred by the rapid decomposition of Rg₁ and Rh₁ by MT619 into F₁ and PPT. Released glucose and rhamnose moieties were also identified in the Re and Rg₂ mixtures (Figure S2).

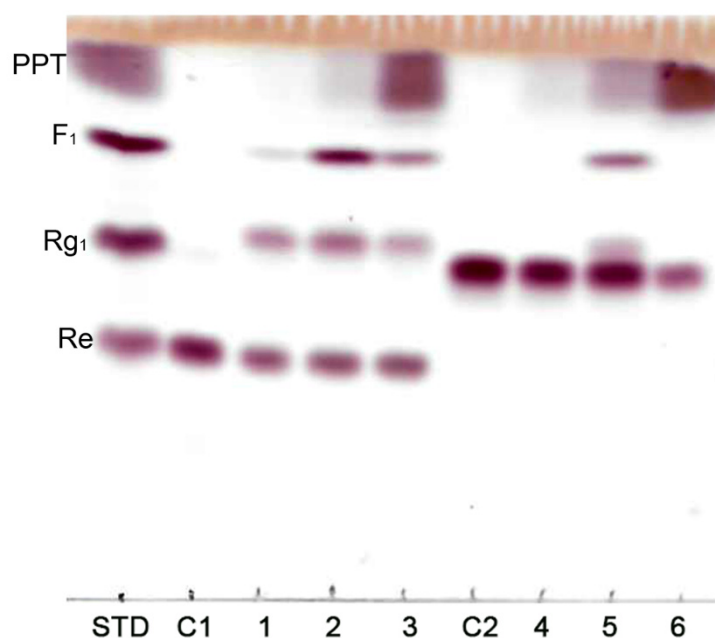


Figure 1. Thin layer chromatography of hydrolyzation products by MT619. C1, Re; 1, Re transformation products; 2, Re transformation products with F₁ added; 3, Re transformation products with PPT added; C2, Rg₂(S); 4, Rg₂(S) transformation products; 5, Rg₂(S) transformation products with F₁ added; 6, Rg₂(S) transformation products with PPT added.

3.2. Optimization for Ginsenoside MT1 Production

Optimal conditions for the biotransformation of Re into MT1 were determined by examining the substrate concentrations in the reaction mixture. Enzyme reactions were performed using of ginsenoside Re, which is one of the major ginsenosides in ginseng [23,24]. The concentration of MT1 increased in proportion to the substrate concentration (Figure 2A); the maximum concentration of MT1 produced was 8.0 mg/mL with 22.5 mg/mL Re 24 h after reaction. Note that >41.2% of Re remained after 24 h when the Re concentration was >15.0 mg/mL (Figure 2B); however, nearly all the Re was converted when the Re concentration was 6.0 mg/mL. Therefore 6.0 mg/mL Re was used for the scaled-up production of MT1.

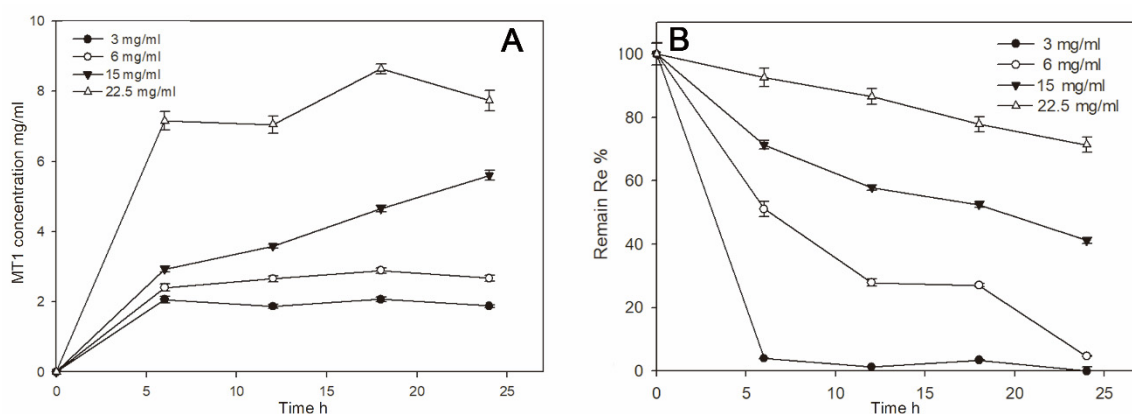


Figure 2. Effects of the substrate Re concentrations (A) and remaining Re concentrations (B) in the reaction mixture.

3.3. Mass Production of MT1

Six grams of Re in 1.0 L of phosphate buffer (pH 7.0) was reacted with cell lysates containing MT619. Re was completely converted 24 h after adding MT619 (Figure 3; Figure 4B). To remove proteins and impurities, the supernatants of the reaction mixture were applied to a 120 g HP20 macroporous resin. After washing with water, ethanol was used to elute the ginsenosides from the resin. The eluate was then evaporated in vacuo, yielding 3.8 g of mixed dry ginsenosides. In preparation for further purification, 1.0 g MT1 was dissolved in 200 mL of 10% methanol; the undissolved precipitants were separated using an ODS column to yield 13 fractions (Figure S3). MT1 was eluted in the 70% methanol fractions (9 and 10), yielding 655 mg of dry compounds.

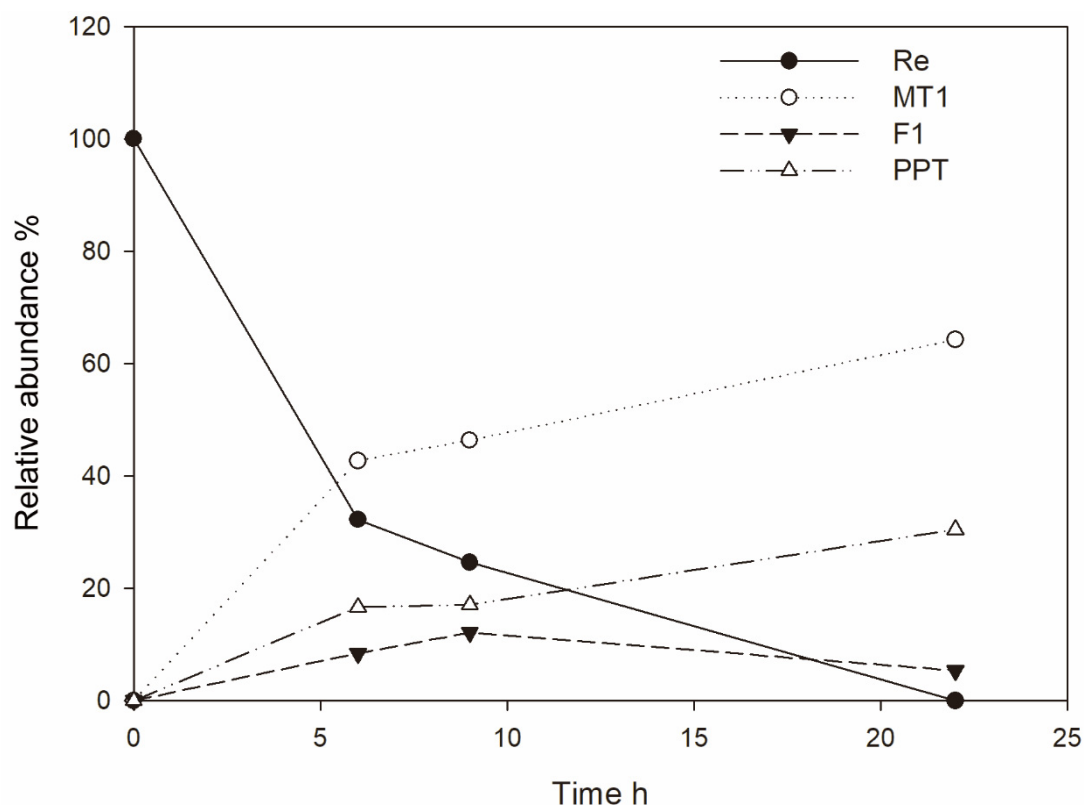


Figure 3. The relative abundance of ginsenosides Re, MT1, F1 and PPT in a scaled-up production reactor over time.

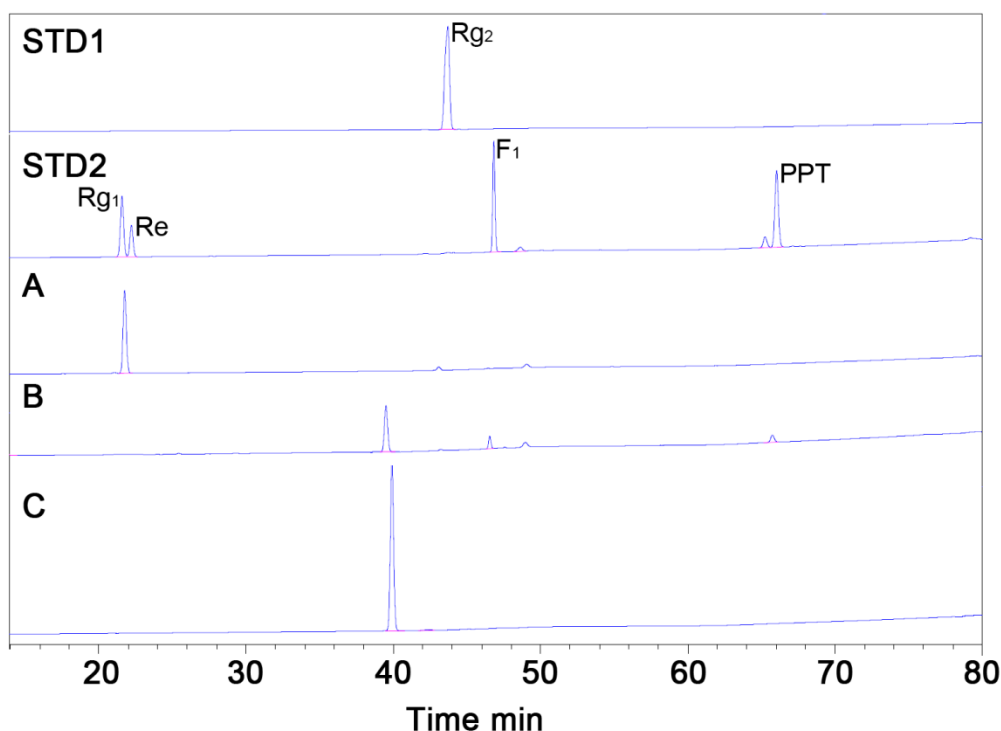


Figure 4. HPLC analysis of purified MT1 from the product mixture. (STD1), ginsenoside Rg₂(S) standard; (STD2), standards of ginsenosides Rg₁, Re, F₁ and PPT; (A), substrate ginsenoside Re before purification; (B), biotransformed products from the MT619 conversion mixture; (C), MT1 isolated using Recycling preparative high-performance liquid chromatography (RPHPLC).

3.4. Purification of MT1 using RPHPLC

Recycling preparative HPLC, which can enhance the separation of compounds by recycling the effluent sample many times over the column without increasing the length of the chromatographic column, was used for further purification. MT1 was purified from the product mixture with a preparative ODS column by RPHPLC. Dry MT1 mixture (300 mg) was loaded onto the RPHPLC column for separation. MT1 was baseline-resolved after three effective columns (Figure S4). The purification process resulted in 214 mg of MT1 with 98.9% chromatographic purity (Figure 4C). The total yield of the production of MT1 from Re was 47.4%.

3.5. Structural Characterization of Ginsenoside MT1

Pure ginsenoside MT1 was obtained as a white powder with a molecular ion peak at m/z 785.5 in the positive LC-MS, which corresponds to the molecular formula C₄₂H₇₂O₁₃ (Figure S5). Acidic hydrolysis of the compound yielded neohesperidose, the signals of which were similar to those of other neohesperidose-containing compounds, such as ginsenoside Rg₂(S) and neohesperidin (Figure S6).

Structural characterization was performed by a Bruker AVANCE III 700 NMR spectrometer operating at 700 MHz (¹H) and 175 MHz (¹³C) with chemical shifts given in ppm for atomic force microscopy (AFM) to determine the chemical structure of MT1 (Table 1). The ¹H-NMR spectrum showed one olefinic (δ H 5.28), two anomeric (δ 5.25 and 6.61) protons and a methyl proton signal (δ 1.83) like in L-rhamnopyranoside. Coupling constants suggested the configuration of the anomeric positions to be α -and- β form for the anomeric proton signals in the ¹H-NMR spectrum. The ¹³C-NMR signals of the sugar were observed as one hemiacetal (δ C 96.7), four oxygenated methines (δ C 79.8, 78.4, 76.6 and 71.6) and one oxygenated methylene (δ C 62.95), indicating that the sugar is β -glucopyranose.

Table 1. ^1H - (700 MHz) and ^{13}C -NMR (175 MHz) spectra of MT1.

Aglycon Moiety Position	δ_{H} mult., (J Hz)	δ_{C} mult.	Sugar Moiety Position	δ_{H} mult., (J Hz)	δ_{C} mult.
1	1.73 m, 1.06 m	39.4 t	Glc-1'	5.25 d (6.3)	96.7 d
2	1.93 m, 1.88 m	28.2 t	Glc-2'	4.29 ^b	76.6 d
3	3.56 t-like (5.6)	78.4 d	Glc-3'	4.29 ^b	79.8 d
4		40.4 s	Glc-4'	4.09 t (9.1)	71.6 d
5	1.27 d (10.5)	61.8 d	Glc-5'	3.88 m	78.4 d
6	4.40 ^c	67.8 d	Glc-6'	4.40 ^c , 4.25 m	62.5 t
7	2.04 m, 1.90 m	47.5 t	Rha-1''	6.61 s	101.4 d
8		41.2 s	Rha-2''	4.79 brs	72.5 d
9	1.67 m	49.7 d	Rha-3''	4.64 t (4.9)	72.6 d
10		39.3 s	Rha-4''	4.40 ^c	74.2 d
11	2.20 m, 1.55 m	31.1 t	Rha-5''	4.89 dd (9.1, 5.6)	69.4 d
12	4.11 m	70.7 d	Rha-6''	1.83 d (5.6)	19.0 q
13	2.00 m	49.0 d	3'-OH	7.50 brs	
14		51.6 s	4'-OH	7.40 d (4.2)	
15	1.58 m, 1.03 m	30.9 t	6'-OH	5.90 brs	
16	1.95 m, 1.44 m	26.7 t	2''-OH	6.76 d (4.2)	
17	2.77 dd (18.2, 10.5)	53.3 d	3''-OH	6.46 brs	
18	1.12 s	17.4 q	4''-OH	6.81 d (3.5)	
19	1.05 s	17.5 q			
20		84.0 s			
21	1.59 s	22.8 q			
22	2.45 m, 1.98 m	35.9 t			
23	2.29 m	24.0 t			
24	5.28 ^d	126.0 d			
25		130.8 s			
26	1.62 s	25.8 q			
27	1.65 s	17.9 q			
28	2.02 s	32.0 q			
29	1.48 s	16.5 q			
30	1.17 s	17.2 q			
3-OH	5.76 d (5.6)				
6-OH	5.28 ^d				
12-OH	5.58 s				

^a Measured at 700 and 175 MHz; obtained in C5D5N with TMS as an internal standard. The assignments were based on ^1H - ^1H COSY, HSQC, and HMBC experiments. ^{b-d} Overlapped with other signals.

The β -D-glucopyranosyl anomeric proton signals were confirmed to be linked at the C-20 position by long-range heteronuclear multiple bond connectivity (HMBC) correlations between the proton signal at δ 5.25 (H-1'') and the carbon signal at δ 84.0 (C-20). A large downfield shift in the C-2'' (δ 76.6) was observed for the inner β -D-glucopyranosyl moiety at C-20 of the aglycone, which showed that the C-1''' (δ 6.61) in the terminal α -L-rhamnopyranosyl moiety is linked to the inner β -D-glucopyranosyl moiety at C-20 [25]. In summary, these results indicate that the metabolite of MT619 is (20S)-3 β ,6 α ,12 β ,20-tetrahydroxydammarane-20-O- $[\alpha$ -L-rhamnopyranosyl(1 \rightarrow 2)]- β -D-glucopyranoside] (Figure 5) or ginsenoside MT1.

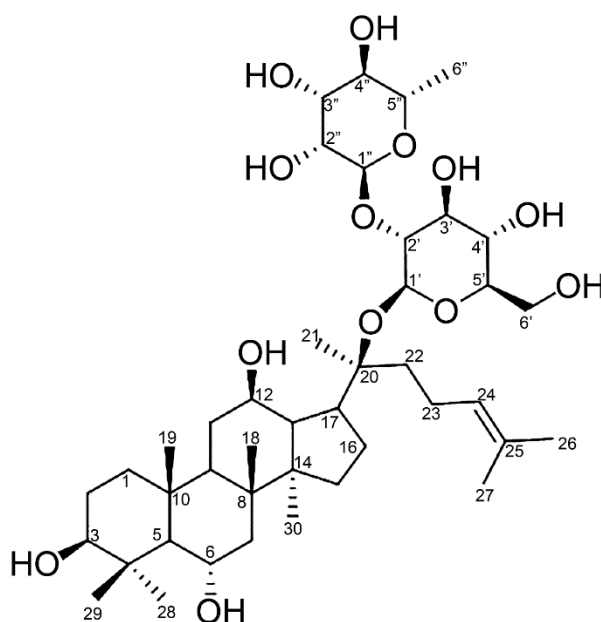


Figure 5. The structure ginsenoside MT1 based on the spectral data.

Based on the above experimental results, the conversion pathways were deduced as shown in Figure 6. In the conversion of Re, the rhamnopyranoside moieties were released from the substrates and transformed into the C-20 outer glucopyranoside of ginsenoside F₁. When MT619 reacted with Rg₂, the rhamnosyl transformation could only occur with the existence of the attached glucopyranoside at C-20 positions such as F₁. The mechanisms of rhamnose transformation by MT619 warrant further exploration.

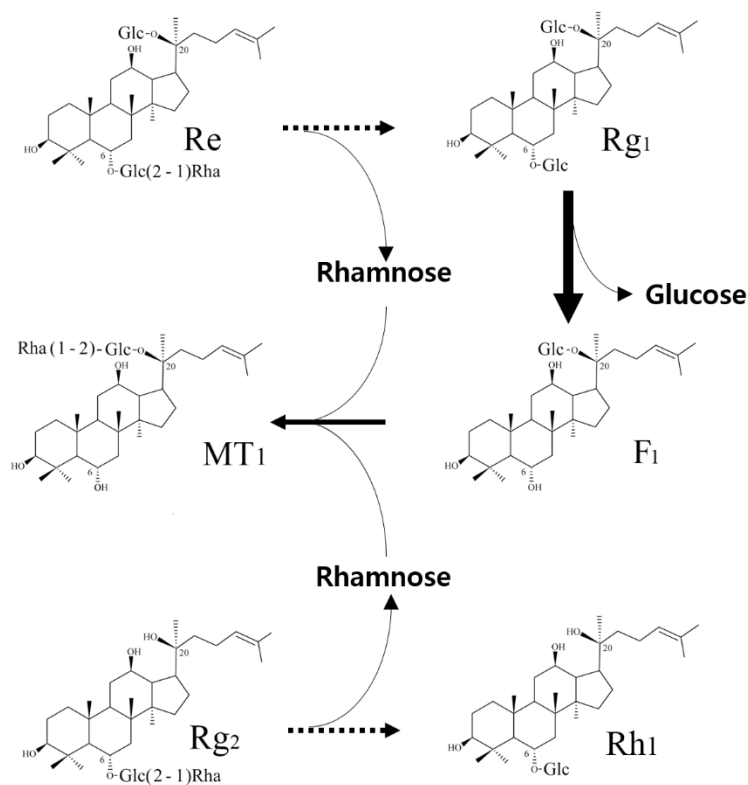


Figure 6. Conversion pathways of ginsenosides Re and Rg₂(S) by MT619.

4. Discussion

Many glycoside hydrolases have been shown to possess transglycosylation capabilities in addition to hydrolytic activity [26]. In substrate transglycosylation of glycoside hydrolases, the glycosyl part of the substrate is transferred to a hydroxyl-containing compound instead of water moieties [27]. Some GH family 3 enzymes also show transglycosylation capabilities from various substrates. A GH3 cellobiase from *Phanerochaete chrysosporium* can hydrolyze and transglycosylate glucopyranoside from laminarioligosaccharides [28]. β -glucosidase from *Aspergillus niger* strain ASKU28 also showed transglucosidase activity against various substrates [29]. MT619 has been classified into GH family 3 according to the Carbohydrate-Active enZymes (CAZy). To the best of our knowledge, this is the only transrhamnosylation activity reported in glycoside hydrolase family 3 to date. Further studies are needed to reveal its sequence-activity relevance.

Some glycosylated ginsenoside derivatives from various natural ginsenosides have been synthesized using enzymatic methods. A novel α -glycosylated ginsenoside F₁((20S)-3 β ,6 α ,12 β -trihydroxydammar-24-ene-(20-O- β -D-glucopyranosyl-(1 \rightarrow 2)- α -D-glucopyranoside)) was synthesized using a cyclodextrin glucanotransferase [30]; ginsenoside Ia((20S)-3b,6a,12b,20-tetrahydroxydammar-24-ene-20-O-b-D-glucopyranosyl-3-O-b-D-glucopyranoside) was synthesized using UDP- glycosyl transferase (BSGT1) from ginsenoside F₁ [31]. In addition, various Rg₁ α -glycosylated ginsenosides (α -1,3-, α -1,4- or α -1,6-glucosidic linkage in β -glucose moieties) linked at C-6 and C-20 positions of aglycones have been synthesized using rice seed α -glucosidases [32], although no trans-rhamnosyl activity has been found using ginsenosides as rhamnose receptors.

A similar ginsenoside—ginsenoside Rg₁₈ (Figure S1), which differs from MT1 in that it is linked to an additional glucose moiety at the C-6 position—was recently found, isolated and characterized from *Panax ginseng* roots [33]. Rg₁₈ has been shown to have various important pharmacological effects, such as anticancer [34] and neuroprotective effects [35]. However, the content of Rg₁₈ in ginseng is <36.0 μ g/g of ginseng, which means that its application is limited by its scarcity in plants. Fortunately, MT1 can be produced efficiently from the relatively abundant ginsenoside Re using the recombinant MT619 with high recovery, which can make it industrially applicable. Furthermore, the deglycosylated ginsenosides generally have shown enhanced pharmacological effects and absorbances than the original ginsenosides [36–38]. Our group is drawing on the properties of MT1 to conduct experiments on compounds with higher anticancer, anti-inflammatory, anti-oxidation and anti-aging.

5. Conclusions

In summary, this paper describes characterization and mass production of a new dammarane-type triterpene saponin from enzymatic converted products. The trans-rhamnosyl activity of MT619 against ginsenoside Re to F₁ resulted in the new ginsenoside MT1, a PPT-type ginsenoside containing one neohesperidose moiety at the C-20 position (confirmed based on spectral data). This unknown compound can be efficiently produced from abundant ginsenoside Re using MT619. Various studies are on-going by our group to investigate its pharmacological effects. We expect potential commercial uses of MT1 in nutraceutical, pharmaceutical or cosmeceutical areas.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2218-273X/10/4/525/s1>, Figure S1: Chemical structures of ginsenosides; Figure S2. Released glucose (orange arrow) and rhamnose (blue arrows) in the supernatant of the reaction products of Re and Rg₂(S); Figure S3. Thin layer analysis of the fractions for the purification of MT1 using the ODS column; Figure S4. RPHPLC chromatogram showing the resolution of the Rb3 and Rd; Figure S5. Positive-ion mode HPLC-ESI-MS/MS of the molecular ion (sodium adduct) of MT1; Figure S6. TLC analysis of acidic hydrolysis products from neohesperidose-moiety harboring substrates by hydrogen chloride.

Author Contributions: Conceptualization, C.-h.C., S.-C.K. and B.-M.J.; methodology, C.-h.C., B.-M.J., S.-C.K. and J.-I.B.; software, C.-h.C., B.-M.J., S.-C.K. and J.-I.B.; validation, C.-h.C., J.-I.B., S.-C.K. and B.-M.J. formal analysis, B.-M.J. and J.-I.B.; investigation, C.-h.C., J.-I.B., M.K. and S.-C.K.; resources, C.-h.C., B.-M.J. and J.-I.B.; data curation, C.-h.C., J.-I.B., S.-C.K. and B.-M.J.; writing — original draft preparation, C.-h.C. and J.-I.B.; writing — review & editing, J.-I.B., M.-S.K., S.-C.K. and B.-M.J.; Visualization, C.-h.C. and J.-I.B.; supervision, S.-C.K.;

project administration, S.-C.K. and C.-h.C.; funding acquisition, S.-C.K.. All authors have read and agreed to the published version of the manuscript.

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