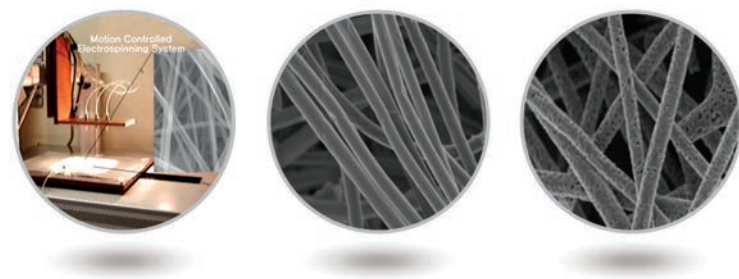




Ten Achievements of 2013 that put KAIST in the Spotlight

High-capacity and Long-cycle Lithium-Oxygen Batteries

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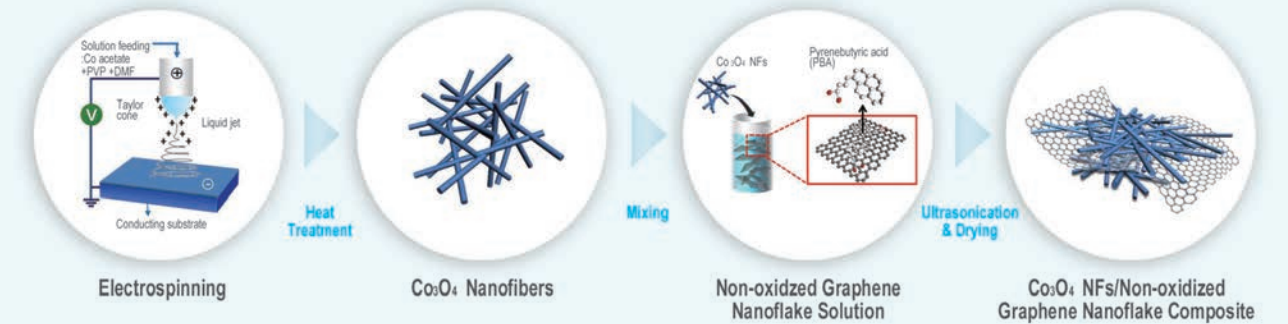


Synthesis of Nanofiber Catalysts and High Performance Lithium-Oxygen Batteries

To solve global issues regarding shortages of fossil fuels and greenhouse effects due to abrupt carbon dioxide emission, the development of eco-friendly, next-generation, electric vehicles has been drastically accelerated. In 2030, lithium-oxygen batteries using high performance and cost-effective catalysts will have been successfully developed and electric vehicles composed of new lithium-oxygen battery systems will be able to make 800 km round trips without recharging. By significantly reducing the purchase cost of electric vehicles, such vehicles will gradually spread to the general public, thereby relieving more and more environmental issues such as abnormal climate change.

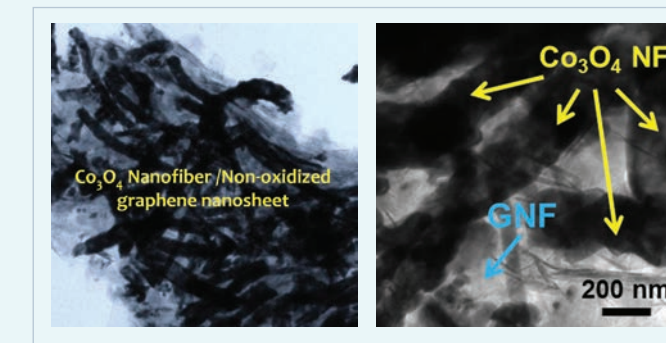
Lithium-oxygen (Li-O₂) batteries, which involve a unique operating mechanism ($2\text{Li} + \text{O}_2 \leftrightarrow \text{Li}_2\text{O}_2$) related to the formation and decomposition of solid lithium peroxide (Li₂O₂) products onto carbonaceous electrodes, have received a lot of attention due to their energy density that is five to ten times higher than that of Li-ion batteries. However, Li-O₂ batteries often suffer from poor performance due to the low electron conductivity of the insulating Li₂O₂ products that accumulate on the surface of the electrodes after discharging. Therefore, current Li-O₂ batteries have several major problems such as low efficiency, extremely poor cycling, and limited rate capability, and these issues should be resolved so that these batteries can be used for impractical applications. Many studies have thus focused on developing highly efficient and cost-effective catalysts for O₂ cathodes in order to significantly improve the reaction kinetics of Li-O₂ batteries.

This study focused on the development of high aspect ratio, one-dimensional (1-D), Co₃O₄ nanofibers attached to 2-D non-oxidized graphene nanoflakes as a highly efficient catalyst of an oxygen electrode for high capacity and long cycle Li-O₂ batteries. In tailored hybrid nanocomposite catalysts, 1-D Co₃O₄ nanofibers provide a high degree of catalytic activity for both the oxygen

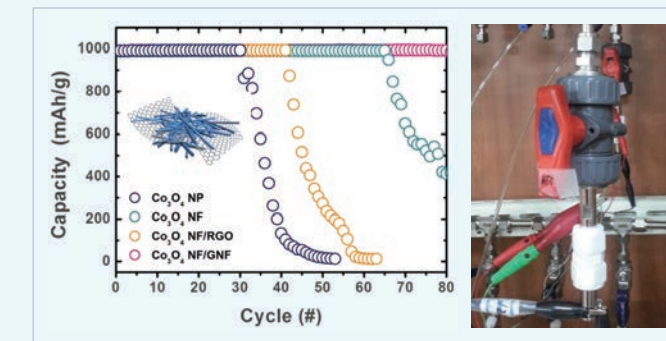


(Figure 1) Schematic illustration of the synthetic strategy of Co₃O₄ NF/graphene composite

evolving reaction and the oxygen reduction reaction; 2-D non-oxidized graphene nanoflakes offer high surface area and excellent electronic conductivity. In this regard, Li-O₂ batteries that are composed of hybrid nanocomposite catalysts delivered a high 1st discharge capacity of 10,500 mAh/g and superior cyclability for 80 cycles with a limited capacity of 1,000 mAh/g. The 1-D Co₃O₄ nanofibers attached to 2-D non-oxidized graphene nanoflakes exhibited superb performance as highly efficient catalysts for high capacity and long cycle Li-O₂ batteries; this performance is in contrast to that of previously reported nano catalysts.



(Figure 2) Morphologies of Co₃O₄ NF/GNF composite



(Figure 3) Cycling data of Co₃O₄ NF/graphene composite and photograph of Li-O₂ battery cell

Expectation Effectiveness

The 1-D Co₃O₄ nanofibers attached to 2-D non-oxidized graphene nanoflakes can be introduced as a high performance and cost-effective catalyst for Li-O₂ batteries instead of expensive, existing catalysts, such as platinum or gold. The Li-O₂ battery system, which is optimized for next generation electric vehicles will enable us to drive a 800 km round trip without recharging. Furthermore, electrospinning as a most suitable synthetic route for mass-production of 1-D structures, can be successfully applied to fabricate diverse classes of nanofiber catalysts for Li-O₂ batteries.

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Research Results

Patents Pending: one patent
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