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Tailoring the Mesoscopic TiO₂ Layer: Concomitant Parameters for Enabling High-Performance Perovskite Solar Cells

Taehyun Hwang^{1†}, Sangheon Lee^{1†}, Jinhyun Kim¹, Jaewon Kim¹, Chunjoong Kim², Byungha Shin³ and Byungwoo Park^{1*}

Abstract

Architectural control over the mesoporous TiO₂ film, a common electron-transport layer for organic-inorganic hybrid perovskite solar cells, is conducted by employing sub-micron sized polystyrene beads as sacrificial template. Such tailored TiO₂ layer is shown to induce asymmetric enhancement of light absorption notably in the long-wavelength region with red-shifted absorption onset of perovskite, leading to ~20% increase of photocurrent and ~10% increase of power conversion efficiency. This enhancement is likely to be originated from the enlarged CH₃NH₃Pbl₃(Cl) grains residing in the sub-micron pores rather than from the effect of reduced perovskite-TiO₂ interfacial area, which is supported from optical bandgap change, haze transmission of incident light, and one-diode model parameters correlated with the internal surface area of microporous TiO₂ layers. With the templating strategy suggested, the necessity of proper hole-blocking method is discussed to prevent any direct contact of the large perovskite grains infiltrated into the intended pores of TiO₂ scaffold, further mitigating the interfacial recombination and leading to ~20% improvement in power conversion efficiency compared with the control device using conventional solution-processed hole blocking TiO₂. Thereby, the imperatives that originate from the structural engineering of the electron-transport layer are discussed to understand the governing elements for the improved device performance.

Keywords: Perovskite solar cell, Nanostructural engineering, Light management, Modeling, Shunting path

Background

Organic-inorganic hybrid perovskites (CH₃NH₃PbI₃) have drawn enormous attentions due to their superior optoelectronic properties and versatilities in applications [1, 2]. For photovoltaic devices, many strategies have been attempted to improve the power-conversion efficiency. One among many deals with refining the perovskite film itself to reduce the trap states and unwanted electron-hole recombination. Generally, defects in grains or grain boundaries act as trap sites for the charge carriers and consequently decrease the charge collection efficiency [3–6]. Indeed, much effort aimed at the

single-crystal perovskites caused successful results for the high photon-to-charge conversion efficiency [7–10]. Therefore, examining the strategies to control the crystallization for the defect reduction is necessary to achieve better-performing perovskite photovoltaics.

Defect-reduced perovskite films can be realized by directly modifying the perovskite synthesis conditions (e.g., reaction environment, precursor stoichiometry, crystallization atmosphere, etc.) [11–13] or by altering the mesoscopic structure of the underlying layers over which the perovskite film synthesis is conducted. The perovskite films are generally deposited upon mesoscopic scaffolds composed of oxide nanoparticles like ${\rm TiO_2}$, into which the perovskite precursors infiltrate and form small crystallites whose dimensions are defined by the internal pore size of mother scaffold. Enlarging the pores in the scaffold, and hence, increasing the infiltrated perovskite grains is expected to reduce the defects

Full list of author information is available at the end of the article



^{*} Correspondence: byungwoo@snu.ac.kr

[†]Equal contributors

¹Department of Materials Science and Engineering, WCU Hybrid Materials Program, Research Institute of Advanced Materials, Seoul National University, Seoul 08826, Korea

by grain boundaries. At the same time, the internal electric field that is formed at the semiconductor junction may further assist the charge separation. Light trapping by the nanostructural engineering will also yield an additional merit for the performance enhancement [14].

To exploit the potential benefits of large-sized single crystalline perovskite, we herein controlled the nanostructures of mesoscopic TiO₂ layer to infiltrate the enlarged CH₃NH₃PbI₃(Cl) grains. Introduction of sacrificial templates during photoelectrode fabrication, one of the facile methods to obtain the controlled pore size and internal surface area [15–18], was applied to render submicron sized pores where the large perovskite grains can be accommodated. The concomitant effect of perovskite crystallinity, perovskite-TiO₂ interfacial area, and light trapping was investigated to understand the change of photovoltaic parameters resulted from the templating method. Furthermore, since the templated porous layer with hundred-nanometer large open pores inevitably raises the necessity for the complete compactness of hole-blocking layers against charge recombination at the FTO-perovskite direct contact, an alternative blocking layer was applied, providing additional power-conversion efficiency improvement. The essential issues in nanostructural engineering were discussed with the correlated solar-cell parameters.

Methods

Preparation of Polystyrene (PS)-TiO₂ Mixture Solution

The PS-TiO $_2$ mixture solution was prepared by mixing the ethanol-based PS solution (PS microsphere with 200 nm in diameter) and the TiO $_2$ paste (anatase-phase TiO $_2$ nanoparticles with ~20 nm in diameter) with various ratios (PS/TiO $_2$ = 1:10, 1:5, and 1:2 in wt. % ratio). The PS-TiO $_2$ solution was then diluted with identical solvent (PS-TiO $_2$ /ethanol = 2:5 in wt. % ratio) for spin-coating. To prepare the bare TiO $_2$ solution without polystyrene for a reference, the TiO $_2$ paste was diluted with anhydrous ethanol to the corresponding wt. % ratio.

CH₃NH₃Pbl₃(Cl) Deposition

The PbI_2 pre-coating was performed following our previous report [19]. The 3:1 molar ratio of MAI/PbCl₂ in DMF (perovskite precursor solution; 2.64 M of MAI and 0.88 M of $PbCl_2$) was then spin-coated at 2000 rpm for 60 s on the PbI_2 pre-coated layer or the TiO_2 compact layer and annealed at 100 °C for 50 min. To enhance the coverage and obtain the similar thicknesses of the perovskite capping layers in the PS-templated TiO_2 cases, spin-coating conditions were optimized. Spin-coating speed was reduced from 6500 to 1500 rpm for PbI_2 and from 2000 to 1500 rpm for the perovskite precursor solution. The precursor concentration was increased (from molar ratio of 2.64:0.88 to 4.08:1.36 between MAI and

PbCl₂) with the increased annealing time (from 50 to 135 min), and the optimization was checked in the aspect of the perovskite crystallization from diffraction. As a control group, molar ratio of 2.64:0.88 between MAI and PbCl₂ was also used on the 1:10 PS-templated TiO₂, and we referred it as "1:10 (unoptimized)" since the perovskite did not fully cover the top of the 1:10 PS-templated TiO₂. Every perovskite deposition was processed in air.

Solar Cell Fabrication

A fluorine-doped tin oxide (FTO) substrate was cleaned, and the TiO₂ compact layer was deposited using the 150 and 300 mM solutions of titanium diisopropoxide bis(acetylacetonate) in 1-butanol through the spincoating followed by the annealing at 500 °C [20]. Then, the substrate was immersed in a 40 mM TiCl₄ aqueous solution and treated in 70 °C oven for 30 min, followed by annealing at 500 °C. Bare TiO₂ or PS/TiO2 solution was spin-coated at 2500 rpm for 30 s, and the substrate was annealed at 500 °C to remove the polystyrene templates. Then, TiCl₄ treatment was performed again, and MAPbI₃(Cl) layer was deposited as mentioned in the previous paragraph. Hole transport layer was coated using the spiro-OMeTAD solution (72.8 mg in 1 mL of chlorobenzene) with the addition of 17.5 µL of Li-TFSI stock solution (520 mg in 1 mL of acetonitrile) and 28.8 µL of tert-butylpyridine [20]. Finally, Au electrode was thermally evaporated.

The ${\rm TiO_2}$ compact layer was separately prepared by rf-magnetron sputtering as an alternative blocking layer [21], instead of using conventional titanium diisopropoxide bis(acetylacetonate) solution. The deposition was performed using the ${\rm TiO_2}$ target (anatase, 99.99%; 5-cm diameter and 0.6-cm thickness) at room temperature under an Ar atmosphere with the operating pressure of 13 mTorr and rf power of 120 W. Except for the blocking layer deposition, all the other procedures were exactly identical to the solar cell fabrication conditions described above.

Characterization

The crystal structure was examined by X-ray diffraction (XRD) (D8 Advance: Bruker). The images of secondary electrons and back scattered electrons were collected from field-emission scanning electron microscope (FESEM) (Merlin Compact: Zeiss), with the energy-dispersive X-ray spectroscopy (SEM-EDS). The absorbance and transmittance of the films were recorded through a UV-Vis spectrophotometer (Cary 5000: Agilent Technologies) with the integrating sphere, and the optical bandgap was evaluated from the α^2 vs. hv (photon energy) analysis. Photocurrent density-voltage (J-V) curves

were obtained by the solar cell measurement system (K3000: McScience) with a solar simulator (Xenon lamp, air mass (AM) 1.5 at 100 mW cm $^{-2}$). During the measurement, black mask of 0.09 cm 2 was applied, and the scan rate was fixed to 150 mV s $^{-1}$ (reverse direction).

Results and Discussion

For the achievement of high photon-to-charge conversion efficiency in solar cell operation, the high light absorption followed by the electron-hole generation and facile separation of carriers into each electrode should be guaranteed throughout the cell structure. Thus, the essential parameters that can affect these phenomena should be considered [22-25]. For high photoresponsivity, the composition and morphology of MAPbI₃ can be altered to broaden the absorption spectra [26-29]. For the electron-hole pair separation, internal electric field driven from the semiconductor junction can be utilized, and it is supported by the result that the MAPbI₃ phase forms the depletion region at the interface with TiO2 in approximately hundreds of nanometers [30]. Having sufficiently large pores in the scaffold, the size of which is comparable to the depletion layer in the perovskite, therefore shall give a microstructural modification of infiltrated perovskite grains with the size desirable in terms of electron-hole separation. A comprehensive outline for the approach suggested above is given in Fig. 1, depicting the nanostructural engineering of ${\rm TiO_2}$ accompanying the perovskite deposition for the intended large crystal infiltration.

To amend the pore size of TiO₂ layers and finally to adjust the grain size of infiltrated perovskite, a sacrificial template is facilely incorporated into the commercially available nanoparticle-based TiO2 pastes by mixing with sub-micron sized polystyrene (PS) beads, varying the composition from $PS/TiO_2 = 1:10$ to 1:2 [15–18]. Rather thick TiO2 porous film (~800 nm) is used for solar cell in this case to investigate the effects of interface between perovskite and TiO₂ on the cell performance [2, 22]. Micropores left after the PS removal are successfully filled with PbI2 by the pre-coating step, and the remaining PbI2 crystals are stacked on TiO2 (Fig. 2a-d and Additional file 1: Figure S1). These pre-coating method guarantees the enlarged grains and crystallinity of the converted perovskite since the original PbI₆ octahedron in the PbI₂ structure maintains its framework after the reaction with MA⁺ and I⁻ in the precursor [19]. As shown in Fig. 2e and Additional file 1: Figure S1, layered-PbI₂ crystals are converted into perovskite, filling the intended ~200-nm micropores. Also, the conversion into MAPbI₃(Cl) is completed while maintaining the [110] orientation without remnant, as verified from the diffraction in Fig. 3a (magnification in Additional file 1: Figure S2(a)).

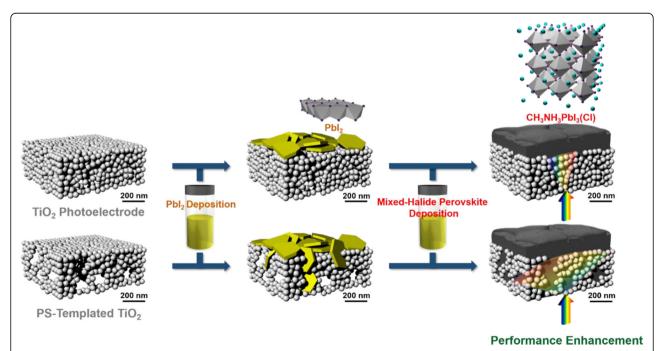


Fig. 1 Schematic diagram illustrating the deposition of mixed-halide perovskite on the nanostructure-tailored TiO₂ scaffold. *Upper row* is the MAPbl₃(Cl) deposition on the general mesoporous TiO₂ substrate. *Lower* is similar except for the nanostructural engineering of TiO₂ using polystyrene (PS) as a sacrificial template. *Yellow crystals* are Pbl₂ consisting of the edge-sharing of Pbl₆ octahedrons, and *black crystals* are mixed-halide perovskites consisting of corner-sharing of Pbl₆ with MA⁺ insertion (*light-blue*)

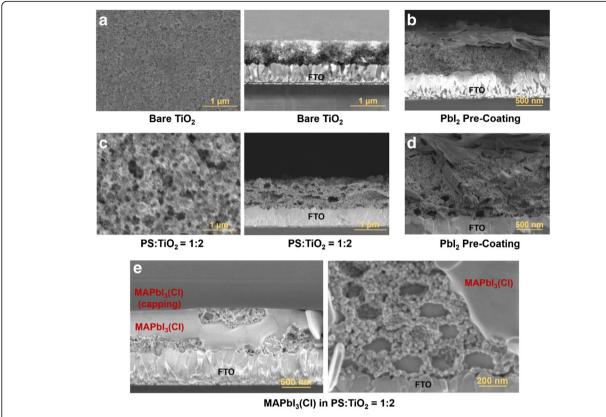


Fig. 2 Scanning electron microscopy images showing the TiO_2 nanostructures with the Pbl_2 pre-coating and $MAPbl_3(Cl)$ infiltration into the polystyrene-templated TiO_2 scaffold. **a** Plan and cross-sectional view of bare TiO_2 . **b** Pbl_2 -pre-coated bare TiO_2 . **c** Plan and cross-sectional view of TiO_2 made from the 1:2 wt. % ratio of $PS/TiO_2 = 1:2$). **d** Pbl_2 -pre-coated TiO_2 from $PS/TiO_2 = 1:2$. **e** Cross section of $PS/TiO_2 = 1:2$ and the magnified view

The back-scattered electron (BSE) imaging is a useful tool to identify the compositional contrast which originates from the atomic-number difference [31]. The BSE images in Additional file 1: Figure S3 confirm that regular ellipsoidal perovskites are clearly formed in the intended micropores. Furthermore, it is used to confirm the PbI₂ pre-coating influence on the perovskite infiltration into the mesoporous TiO₂ layer (mp-TiO₂) [32]. The PbI₂ pre-coating indeed do not interfere with the perovskite infiltration into the mp-TiO₂ (without PS) based on the BSE intensity comparison between Additional file 1: Figure S3(b) and (c). This is further examined by the elemental mapping (SEM-EDS): the distributions of Pb and I are the same whether the PbI₂ pre-coating is performed or not (Additional file 1: Figure S4(a) and (b)) and whether the TiO₂ layer is altered by the PS sacrificial template or not (Additional file 1: Figure S4(b) and (c)). The BSE intensity and the EDS mapping confirm that the interfacial area between the perovskite and TiO2 is decreased with the increased PS fraction, since the nanoparticulated-TiO₂ film consisting of ~20 nm-sized-nanoparticle has a larger internal surface than the ${\rm TiO_2}$ film with the intended ~200-nm micropores. The enlarged perovskite grain by PS incorporation is supported accordingly from the above results.

Haze transmission is the ratio of the diffused transmittance to the total transmittance, and discloses the degree of incident light scattering [33]. The PStemplated TiO2 looks opaque, and the haze increases as the PS ratio rises (Fig. 3b). Also, asymmetric elevation of absorbance is observed from MAPbI₃(Cl) with increasing PS ratio as shown in Fig. 3c. This is due to the increased light scattering from TiO₂ and perovskite by the intended large crystals. In addition, the bandgap of mixed-halide perovskite is red-shifted by ~10 meV from the Tauc plot (Fig. 3d). This optical bandgap change is also observed when the identical mixed-halide precursor solution is used for the bare and 1:10 cases ("unoptimized" which is explained in the experimental section). This red-shift is not from the different quantity of Cl since the (110) peak of MAPbI₃(Cl) is identical between the bare and 1:10 case (Additional file 1: Figure S2(b)) [34].

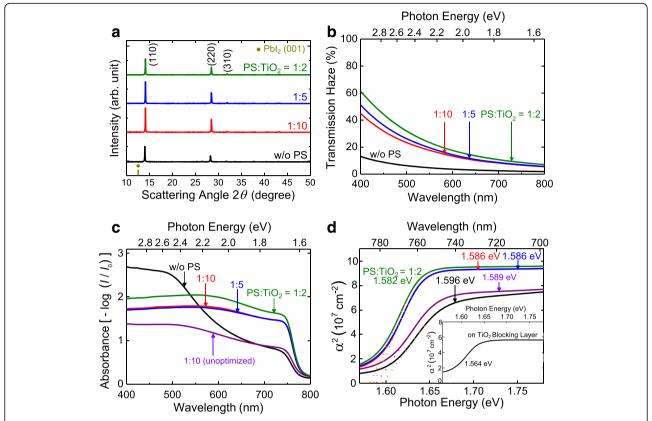


Fig. 3 The effect of PS ratio on the nanostructures and optical properties of MAPbl₃(Cl) perovskite. **a** X-ray diffraction of MAPbl₃(Cl) films. **b** Transmission haze of TiO_2 with different ratios of PS bead. **c** Absorbance of MAPbl₃(Cl) on the corresponding TiO_2 , and **d** the determination of the optical bandgap for the MAPbl₃(Cl) film. Samples are without PS bead (w/o PS), 1:10 wt. % ratio of PS bead in TiO_2 paste (PS/ TiO_2 = 1:10 (before and after optimization)), 1:5 wt. % ratio of PS to TiO_2 (1:5 (optimized)), and 1:2 wt. % ratio (1:2 (optimized)). For comparison, the optical bandgap energy of MAPbl₃(Cl) film on a compact TiO_2 is shown in the *inset* of (**d**)

When the concentration of mixed-halide solution is increased by ~50% while maintaining the MAI/PbCl₂ ratio as 3:1 to improve the perovskite coverage for the PStemplated TiO₂ cases, the (110) peak shifts to the high scattering angle (Additional file 1: Figure S2(a), identical to Fig. 3a with proper magnification). The lattice parameters a and c in tetragonal (space group I4/m) are changed, respectively, from 0.892 to 0.886 nm and from 1.261 to 1.251 nm. The apparent optical bandgap can vary by the Cl concentration in MAPbI₃(Cl), Burstein-Moss effect (carrier concentration), quantum confinement effect, and/or grains and grain boundaries [9, 10, 26, 35-37]. The Burstein-Moss and quantum confinement effects are not pertinent to this system considering that the composition of perovskite was confirmed to be the same for all the cases, and the grain size was out of the regime where the quantum confinement effect works in [36, 37]. Therefore, the optical bandgap change is expected to be caused by the increased perovskite crystal sizes and Cl concentrations (based on the tetragonal unitcell size). To verify the grain-size effect on the absorption shift, MAPbI $_3$ (Cl) perovskite is deposited with the identical concentration of mixed-halide solution to the bare (without PS) on planar ${\rm TiO}_2$ layer (Additional file 1: Figure S5). From the diffraction and SEM images of MAPbI $_3$ (Cl) film, MAPbI $_3$ (Cl) grown on the ${\rm TiO}_2$ planar layer exhibits micrometer-sized lateral grains with the ~30-meV red-shift compared to the bare (without PS) in the absorption onset (inset in Fig. 3d), and this additionally supports the absorption-edge shift with respect to the perovskite grain size.

The MAPbI₃(Cl) solar cells are fabricated on each PS-templated TiO₂ with varying PS ratios (Fig. 4 and Additional file 1: Figure S6) to understand the grain size and interfacial effects on the solar cell performance. The best and the average values of short-circuit current density ($J_{\rm sc}$), open-circuit voltage ($V_{\rm oc}$), fill factor (FF), and power-conversion efficiency (η) are summarized in Table 1. For PS/TiO₂ = 1:10 case, the η is rather decreased in spite of ~13% improvement of $J_{\rm sc}$, which is due to the inferior $V_{\rm oc}$ and FF. Inferior $V_{\rm oc}$ in this case should be resolved to overcome

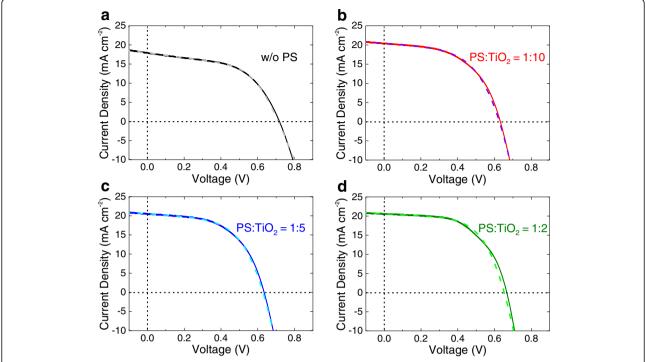


Fig. 4 The effect of PS ratio on the performance of the perovskite solar cell. *J-V* curve (*solid line*) on **a** bare TiO_2 , **b** 1:10, **c** 1:5, and **d** 1:2 PS-templated TiO_2 under light exposure (AM 1.5, 100 mW cm⁻²), and the corresponding fitting result (*dashed line*) from the ideal one-diode model (described in the following Additional file 1: Figure S7). The corresponding photovoltaic parameters are summarized in Tables 1 and 2

the low efficiency, and we have considered several approaches, specifically focusing on the defect reduction that can cause recombination in the perovskite and at the interfaces [38–42]. However, as the PS ratio is increased, these parameters are recovered by ~20 mV and ~4% with the additional increase of $J_{\rm sc}$, leading to approximately 10% increase of η for the PS/TiO₂ = 1:2 case compared with the control devices without PS templating.

Table 1 The effect of polystyrene (PS) ratio on the perovskite solar cell performance

Substrate	$J_{\rm sc}$ [mA cm ⁻²]	V _{oc} [V]	<i>FF</i> [%]	η [%]
w/o PS	17.9	0.719	53.4	6.87
	(16.8)	(0.713)	(52.1)	(6.24)
1:10	20.5	0.633	52.7	6.84
	(18.9)	(0.621)	(50.8)	(5.94)
1:5	20.6	0.635	54.5	7.13
	(19.3)	(0.626)	(53.9)	(6.51)
1:2	20.5	0.667	56.6	7.74
	(19.7)	(0.644)	(54.7)	(6.93)

Short-circuit current density (J_{sc}) , open-circuit voltage (V_{oc}) , fill factor (FF), and power-conversion efficiency (η) of perovskite solar cells, without PS and with various ratios of PS-templated TiO_2 . TiO_2 blocking layer is deposited by solution. (Data in the bracket are the averaged ones from more than four cells in each condition.)

Analyses of *J-V* curves based on the one-diode model provide useful parameters helpful to figure out the interfacial effects. The *J-V* curves are fitted using the ideal one-diode model described in Additional file 1: Figure S7, and the resultant fit curves are shown with the dashed lines in Fig. 4 with the extracted solar-cell parameters in Table 2 [43, 44]. (Fitting results of *J-V* under dark conditions are shown in Additional file 1: Figure S8.) The fit result shows that the dark-saturation

Table 2 Photovoltaic parameters extracted from the ideal onediode model

Substrate	J ₀ [mA cm ⁻²]	J_{ph} [mA cm ⁻²]	R_s [Ω cm ²]	R_{sh} [Ω cm ²]	R_{sh}^{recom} [Ω cm ²]	n
w/o PS	9.39×10^{-6}	18.6	5.90	152.0	152.0	1.94
	(± 3.62×10^{-6})	(± 0.1)	(± 0.19)	(± 0.8)	(± 0.8)	(± 0.05)
1:10	1.92×10^{-2}	20.6	1.69	286.6	287.2	3.53
	(± 0.25×10^{-2})	(± 0.1)	(± 0.11)	(± 5.7)	(± 5.7)	(± 0.07)
1:5	7.84×10^{-3}	20.7	1.95	303.6	304.7	3.13
	(± 1.27 × 10 ⁻³)	(± 0.1)	(± 0.11)	(± 5.8)	(± 5.8)	(± 0.07)
1:2	2.77×10^{-3}	20.7	2.00	440.4	441.8	2.85
	(± 0.67 × 10 ⁻³)	(± 0.1)	(± 0.16)	(± 10.6)	(± 10.7)	(± 0.09)

Dark-saturation current density (J_0) , photogenerated current density (J_{ph}) , series resistance (R_s) , shunt resistance (R_{sh}) , recombination shunt resistance $(R_{sh}^{(corn)})$, and ideality factor (n), respectively, from the cell of the highest efficiency in each condition (AM 1.5 at 100 mW cm⁻²). (Parameters are described in the following Additional file 1: Figure S7.)

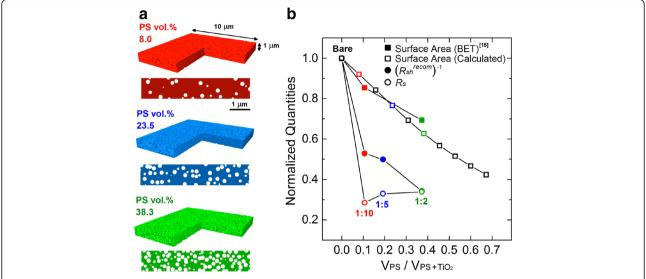


Fig. 5 Calculated surface area for the analyses of surface area of the PS-templated TiO_2 and its comparison with the photovoltaic parameters. **a** Typical simulation results describing the distribution of 200-nm polystyrene-induced pores with the number variation of PS microspheres (PS vol. %) in $10 \times 10 \times 10 \times 10 \times 10$ µm³, and the corresponding typical cross-sectional views. **b** Polystyrene-induced loss of total surface area (*open squares*), plotted with R_s (series resistance) and R_{sh}^{recom} (recombination shunt resistance) from the experimental *J-V* curve, as extracted from the ideal one-diode model. (Simulated surface area is also compared to the experimental values from Ref. [15].)

current density (J_0) and the ideality factor (n) are worsened from $\sim 10^{-5}$ to $\sim 10^{-3}$ mA cm⁻² and ~ 2 to ~ 3 , respectively, when PS is introduced. The back electron transfer from the FTO front electrode to the perovskite by the ~ 200 -nm penetration may cause the recombination path, as seen in SEM image of Fig. 2e [45].

However, the series resistance (R_s) is improved from 5.9 to $2.0~\Omega~{\rm cm}^2$, and the recombination shunt resistance (R_{sh}^{recom}) (reflecting R_{sh} at $J_{ph}=0$) is increased from 152.0 to 441.8 $\Omega~{\rm cm}^2$, which may have led to the enhanced FF. As the PS templating reduces the internal surface area by occupying the internal space for ${\rm TiO_2}$ nanoparticle-based porous structure, improvement of R_s and R_{sh}^{recom} can reasonably be postulated to result from the decrease of interfacial trap sites, which shall be proportional to the internal surface area of ${\rm TiO_2}$ scaffold unless the nature of trap sites at the ${\rm TiO_2}$ -perovskite interface are much affected by PS templating. To check whether such explanation works, the correlation of R_s and R_{sh}^{recom} with the internal surface area of ${\rm TiO_2}$ layer is plotted in Fig. 5.

Table 3 The effect of TiO₂ blocking layer by sputter deposition on the perovskite solar cell performance

Substrate	J _{sc} [mA cm ⁻²]	V _{oc} [V]	<i>FF</i> [%]	η [%]
w/o PS	18.3	0.749	55.2	7.56
1:2	20.8	0.694	58.3	8.41

Short-circuit current density (J_{sc}) , open-circuit voltage (V_{oc}) , fill factor (FF), and power-conversion efficiency (η) of perovskite solar cells from w/o PS and PS/ $TiO_2 = 1:2$. The TiO_2 blocking layer is deposited by sputtering

The internal surface area of PS-mediated TiO₂ layer for various PS fractions is calculated based on a simple Monte-Carlo method [46], and the possible overlap between beads is considered rather than assuming the beads as hard-sphere. The random distribution of 200-nm spheres in $10 \times 10 \times 1$ µm³ volume is simulated by assuming the probability profile of sphere-tosphere overlap to show exponential decay, the exponent of which is assumed following the Hertzian model of elastic potential energy for contact of two identical elastic spheres at a given overlap displacement [47]. A previous report on the surface area change by polystyrene particle templating is also given as a more realistic guidance for comparison. In Fig. 5, the experimental PS/TiO2 weight ratios are converted to volume ratios from the assumed densities of polystyrene (1.05 g cm⁻³), TiO₂ (3.91 g cm⁻³) and porosity of mesoscopic TiO₂ film (68.1%) [15, 48]. By introducing the PS microbeads, the internal surface area is

Table 4 Photovoltaic parameters from the cells with the TiO₂ blocking layer by the sputter deposition

Substrate	J_0 [mA cm ⁻²]	J_{ph} [mA cm ⁻²]	R_s [Ω cm ²]	R_{sh} $[\Omega \text{ cm}^2]$	R_{sh}^{recom} [Ω cm ²]	n
w/o PS	3.53×10^{-7} (± 2.93 × 10 ⁻⁷)	19.0 (± 0.1)	9.53 (± 0.36)	266.7 (± 2.5)	266.7 (± 2.5)	1.64 (± 0.08)
1:2	2.55×10^{-3} (± 0.68×10^{-3})		2.18 (± 0.22)	460.9 (± 11.2)	467.8 (± 11.5)	2.98 (± 0.09)

Dark-saturation current density (J_0) , photogenerated current density (J_{ph}) , series resistance (R_s) , shunt resistance (R_{sh}) , recombination shunt resistance (R_{sh}^{ecom}) , and ideality factor (n), respectively (AM 1.5 at 100 mW cm⁻²)

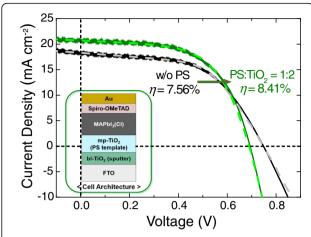


Fig. 6 The effect of TiO_2 blocking layer by sputter deposition on the performance of the perovskite solar cell. *J-V* curve (*solid line*) under illumination with the corresponding fitting results (*dashed line*). Photovoltaic parameters are summarized in Tables 3 and 4

decreased with the improved R_s and $(R_{sh}^{\ recom})^{-1}$. The similar dependence of R_s and $(R_{sh}^{\ recom})^{-1}$ may be from the reduced interfacial traps leading to the decreased resistance of charge transfer or recombination [49]. However, consequently from this simplistic simulation, much drastic variations of R_s and $R_{sh}^{\ recom}$ are correlated with the morphological engineering. This implies that besides the interfacial effects, other factors like the grain size (crystallinity) and thereby the carrier mobility which is affected by the defects or impurities at the grain boundary should be considered to explain the improvement of R_s and $R_{sh}^{\ recom}$, and thereby resultant boost of η [7–9, 50].

To further increase η , blocking layer deposition method is altered from solution deposition (data in

Tables 1 and 2, Fig. 4, and Additional file 1: Figure S6 and S8) to sputter deposition (Tables 3 and 4, Fig. 6, and Additional file 1: Figure S9) to ensure the compactness. Actually, the SEM images in Additional file 1: Figure S10 confirm the porous morphology by the spin-coating and rather compact structure on each FTO grain by sputtering [48, 51-53]. The modified compact blocking layer results in the enhancement of ~ 0.3 mA cm⁻² in J_{sc} , ~ 30 mV in $V_{\rm oc}$ and ~0.7% in η (both bare and PS-templated TiO₂), and consequently, approximately 20% increase of η is achieved through the nanostructural control of both blocking and porous layers compared to the bare sample. The compact TiO₂ hole-blocking layer more effectively inhibits the direct contact between the FTO and the perovskite (preventing the back electron transfer), which is predicted as the origin of charge recombination from the one-diode model. Actually, the modified TiO₂ blocking layer leads to the decreased J_0 (recombination current) in both bare and PS cases with the slight improvement in R_{sh}^{recom} , and these are correlated with the improved V_{oc} and FF as listed in Tables 3 and 4. The performance of the device in this study is not comparable with the state-ofthe-art device due to the low $V_{\rm oc}$, and the hysteresis problem should be resolved. However, the beneficial optical and electrical properties of perovskite are rationally correlated with the nanostructures to elucidate the origin of the enhanced J_{sc} . Furthermore, the principal results from the structural engineering in this work will be applicable for various photovoltaic systems utilizing other metal-oxidebased electron-selective contacts and perovskite compositions due to the simplicity of our approach. In addition, the study applying the defect control will further enhance the $V_{\rm oc}$ in this device architecture which already shows the promising $J_{\rm sc}$ values, potentially improving the device performance even further. As a summary, the effects of

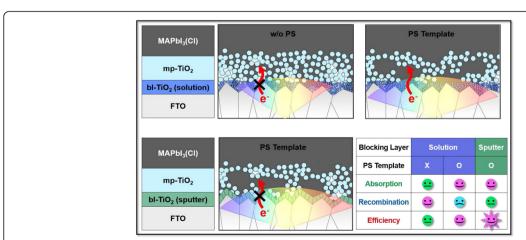


Fig. 7 Schematic illustration demonstrating the roles of PS template and TiO₂ blocking layer on the solar cell performance. The effects of nanostructural engineering by mesoporous TiO₂ (mp-TiO₃) and blocking (bl-TiO₃) layers are illustrated

intended pore engineering in mesoporous ${\rm TiO_2}$ and the blocking layer are illustrated in Fig. 7.

Conclusions

In this work, the mesoscopic TiO₂ structure was facilely engineered by using a sacrificial template, and the perovskite solar cells were fabricated on nanostructurecontrolled scaffolds by systematically examining the concomitant key factors of pore engineering that influence the cell performance. The enhanced efficiency by the enlarged pores was attributed to the effectively infiltrated perovskite grains that provided the beneficial lightharvesting features by the absorption enhancement. The perovskite-TiO2 interfacial area was rationally correlated with the internal resistances of solar cell and associated with the charge transfer and recombination. Consequently, the enlarged perovskite grains with the reduced interfacial area contributed together to the internal resistances, changing direction into the efficiency improvement. The leakage current that caused the recombination was successfully resolved through the compact blocking layer, achieving further performance enhancement. We believe that this work suggests a rational nanostructural design of electron-transport layer for high optoelectronic properties in the emerging solar cells.

Additional file

Additional file 1: Figure S1. SEM images showing the TiO₂ nanostructures with the Pbl₂ pre-coating and MAPbl₃(Cl) infiltration into the PS-templated TiO2. Figure S2. The effect of PS ratio and the concentration of precursor solution on the X-ray diffraction of MAPbl₃(Cl) perovskite. Figure S3. Cross-sectional back scattered electron images exhibiting the MAPbl₃(CI) perovskite infiltration in the porous TiO₂ layer. Figure S4. Cross-sectional elemental distributions from energy dispersive X-ray spectroscopy (SEM-EDS) showing the Sn, Ti, O, Pb, and I distributions for different porous TiO2 scaffolds. Figure S5. Microstructures of MAPbl₃(Cl) on the TiO₂ blocking layer. **Figure S6.** Photovoltaic parameters with the average and the standard deviation in each condition. Figure S7. Ideal one-diode model for the perovskite solar cell. Figure S8. Current density vs. bias under dark and the corresponding fitting results. Figure S9. The effect of TiO₂ blocking layer by sputter deposition on the performance of the perovskite solar cell. Figure S10. Morphology comparison by the spin-coating and sputter deposition.

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Authors' Contributions

TH and SL carried out the overall scientific experiments and drafted the manuscript. JK (Jinhyun Kim) conducted on optimizing the blocking layer deposition. JK (Jaewon Kim) participated in analyzing the X-ray diffraction. CK and BS helped to improve the logical flows in the manuscript. BP supervised this study and gave valuable advices of scientific logics in detail and finalized the manuscript. All authors read and approved the final manuscript.

Competing Interests

The authors declare that they have no competing interests.

Author details

¹Department of Materials Science and Engineering, WCU Hybrid Materials Program, Research Institute of Advanced Materials, Seoul National University, Seoul 08826, Korea. ²School of Materials Science and Engineering, Chungnam National University, Daejeon 34134, Korea. ³Department of Materials Science and Engineering, Korea Advanced Institute of Science and Technology, Daejeon 34141, Korea.

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